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4-Amino-3-methyl-1,2,4-triazole-5-thione derivative of *p*-nitrophenylaldehyde

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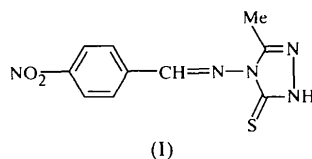
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Abstract

In the title compound, 3-methyl-4-(4-nitrobenzylidene-nitrilo)-4,5-dihydro-1*H*-1,2,4-triazole-5-thione (C₁₀H₉N₅O₂S), the triazole ring is in a planar form. The whole molecule lies in a crystallographic plane. The mean planes through the triazole ring and the phenyl ring form a dihedral angle of 15.9 (1)°. There is an intermolecular N—H···O hydrogen bond.

Comment

Most Schiff bases possess antibacterial, anticancer, anti-inflammatory and antitoxic activities (Williams, 1972), and the sulfur-containing Schiff bases are particularly effective. These Schiff bases are derived from thiosemicarbazone, thiocarbazone and thiocarbohydrazide. Improvements in this biological activity might be achieved by further variation in the chemical structure (Lian *et al.*, 1997; Zhang *et al.*, 1993). We have prepared the title compound, (I), established its crystal structure, and shown that it is highly effective as an inhibitor of *Staphylococcus aureus*.



The bond lengths and bond angles (Table 1) are comparable with reported values (Rodier *et al.*, 1994; Wang *et al.*, 1998). The C1—N1 [1.341 (4) Å] and C1—N3 [1.386 (3) Å] distances are slightly high because of the substitution of the highly electronegative S atom.

The whole molecule is almost planar. The triazole ring of the title compound is planar with a maximum deviation of 0.006 (3) Å for N1. The mean planes through the triazole ring and the phenyl ring form a dihedral angle of 15.9 (1)°. The nitro group is twisted by 9.5 (2)° from the plane of the phenyl ring. The planarity of the molecule is maintained by the intramolecular interactions between C4 and S [3.220 (3) Å] and C9 and O1 [2.706 (5) Å]. Intermolecular N—H···O hydrogen bonds (Table 2) stabilize the packing as well as van der Waals interactions. The molecules and the hydrogen-bonded chains extend in the direction of the *a*-axis.

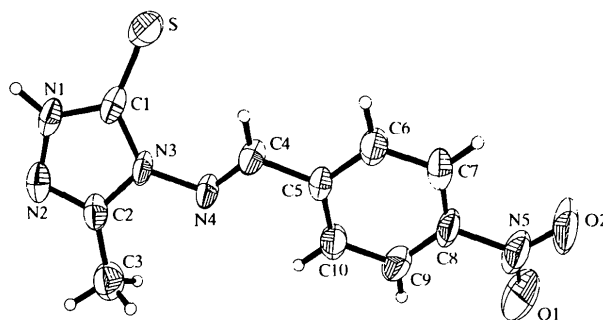


Fig. 1. The structure of the title compound showing 50% probability displacement ellipsoids for non-H atoms and the atom-numbering scheme.

Experimental

4-Amino-3-methyl-1,2,4-triazole-5-thione was synthesized according to the procedure of Mohan (1983) and to a solution of it (0.05 mol, in ethanol 15 ml) was added *p*-nitrophenylaldehyde in ethanol (10 ml). The mixture was acidified to a pH between 4.0–5.0 with hydrochloric acid, then refluxed for 1.5 h at 343 K. An orange solid was recrystallized; yield 70%, m.p. 496.6 K. Single crystals were obtained with great difficulty from acetone by slow evaporation. The crystal used for data collection was a very thin lath, but no better could be obtained.

Crystal data

C₁₀H₉N₅O₂S
M_r = 263.28
 Monoclinic
*P*2₁/*n*
a = 13.3256 (5) Å
b = 6.9775 (3) Å
c = 13.9402 (6) Å
 β = 113.881 (2)°
V = 1185.18 (8) Å³
Z = 4
D_x = 1.476 Mg m⁻³
D_m not measured

Mo K α radiation

λ = 0.71073 Å
 Cell parameters from 2120 reflections
 θ = 2.74–33.22°
 μ = 0.275 mm⁻¹
T = 293 (2) K
 Needle
 0.70 × 0.14 × 0.02 mm
 Yellow

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans

3390 independent reflections
 1525 reflections with $I > 2\sigma(I)$

Absorption correction: $R_{\text{int}} = 0.073$
 empirical (SADABS; $\theta_{\text{max}} = 30.00^\circ$
 Sheldrick, 1996a) $h = -14 \rightarrow 20$
 $T_{\text{min}} = 0.842$, $T_{\text{max}} = 0.996$ $k = -10 \rightarrow 10$
 8482 measured reflections $l = -20 \rightarrow 21$

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0509P)^2$
 $R[F^2 > 2\sigma(F^2)] = 0.077$ $+ 0.2428P]$
 $wR(F^2) = 0.169$ where $P = (F_o^2 + 2F_c^2)/3$
 $S = 1.036$ $(\Delta/\sigma)_{\text{max}} < 0.001$
 3390 reflections $\Delta\rho_{\text{max}} = 0.220 \text{ e } \text{\AA}^{-3}$
 199 parameters $\Delta\rho_{\text{min}} = -0.285 \text{ e } \text{\AA}^{-3}$
 All H-atom parameters Extinction correction: none
 refined Scattering factors from
International Tables for
Crystallography (Vol. C)

Table 1. Selected geometric parameters (\AA , $^\circ$)

S—C1	1.653 (3)	N3—C1	1.386 (3)
O1—N5	1.207 (4)	N3—C2	1.387 (4)
O2—N5	1.212 (4)	N3—N4	1.391 (3)
N1—C1	1.341 (4)	N4—C4	1.261 (4)
N1—N2	1.373 (4)	N5—C8	1.477 (4)
N2—C2	1.302 (4)		
C1—N3—N4	132.7 (2)	N1—C1—S	127.4 (2)
C4—N4—N3	119.5 (2)	N2—C2—C3	125.9 (3)
O1—N5—O2	123.1 (3)	N4—C4—C5	118.2 (3)
O1—N5—C8	118.1 (4)	C7—C8—N5	118.9 (3)
O2—N5—C8	118.8 (4)		

Table 2. Hydrogen-bonding geometry (\AA , $^\circ$)

D—H...A	D—H	H...A	D...A	D—H...A
N1—HN1...O1 ¹	0.88 (4)	2.38 (5)	3.066 (5)	136 (4)
N1—HN1...O2 ¹	0.88 (4)	2.42 (4)	3.290 (4)	170 (4)

Symmetry code: (i) $x - 1, y, z$.

The data collection covered over a hemisphere of reciprocal space by a combination of three sets of exposures; each set had a different φ angle (0, 88 and 180°) for the crystal and each exposure of 30 s covered 0.3° in ω . The crystal-to-detector distance was 4 cm and the detector swing angle was -35° . Coverage of the unique set is over 99% complete. Crystal decay was monitored by repeating thirty initial frames at the end of data collection and analysing the duplicate reflections, and was found to be negligible.

The structure was solved by direct methods and refined by full-matrix least-squares procedures. All H atoms were located from a difference Fourier map and refined isotropically.

Data collection: SMART (Siemens, 1996a). Cell refinement: SAINT (Siemens, 1996b). Data reduction: SAINT. Program(s) used to solve structure: SHELXTL (Sheldrick, 1996b). Program(s) used to refine structure: SHELXTL. Molecular graphics: SHELXTL. Software used to prepare material for publication: SHELXTL and PARST (Nardelli, 1995).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: HA1234). Services for accessing these data are described at the back of the journal.

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2-Hydroxy-5-methyl-3-(morpholinomethyl)-benzaldehyde, (I), and 4,4'-dimethyl-6,6'-bis(morpholinomethyl)-2,2'-ethylenedinitrilodimethyldiphenol, (II)

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Abstract

The title compounds, (I) (C₁₃H₁₇NO₃) and (II) (C₂₈H₃₈N₄O₄), are in the monoclinic space group $P2_1/c$ with $Z = 4$ and 2, respectively. Molecule (II) has a crys-